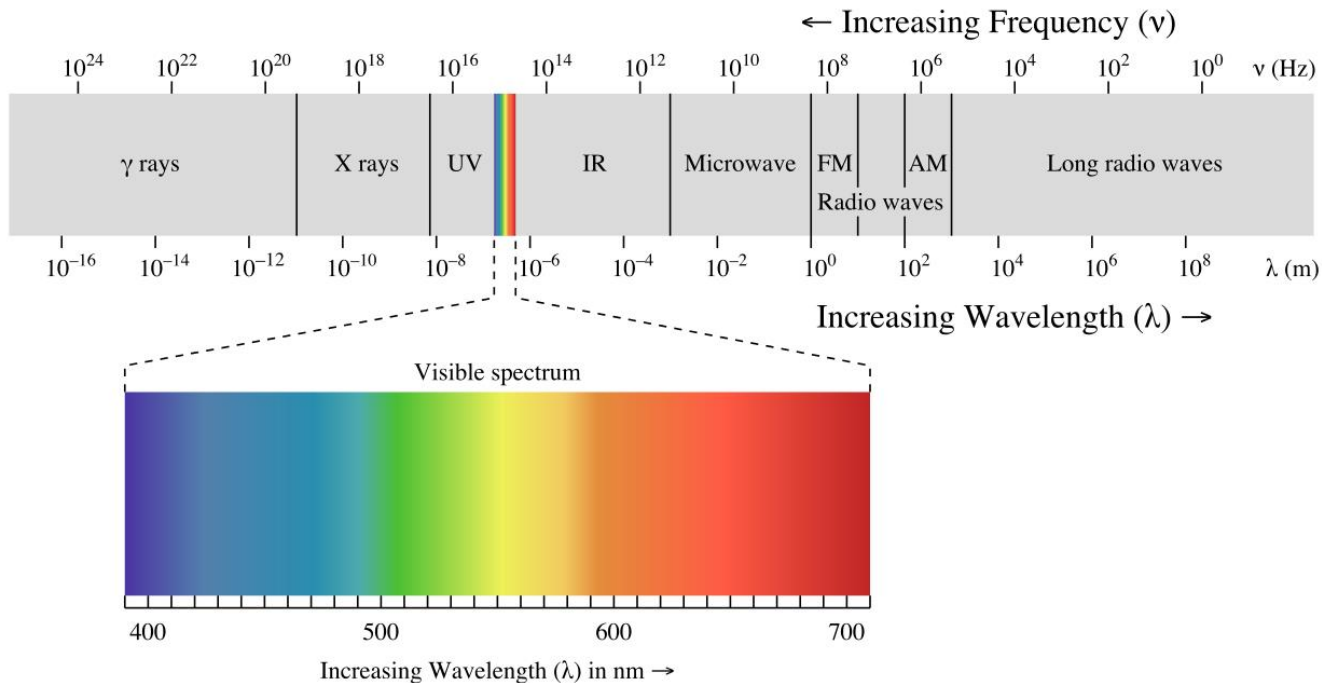


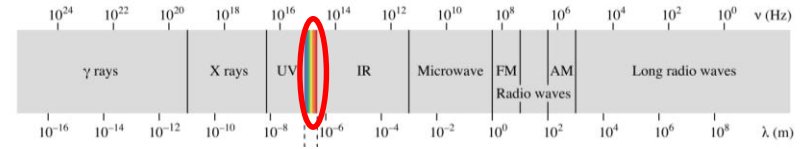
# Optical Spectroscopy:

The study absorption and emission of light in nature

*Abid Khan (past slides from Virginia Lorenz and Kai Wen Teng)*

PHYS 403 Summer 2020





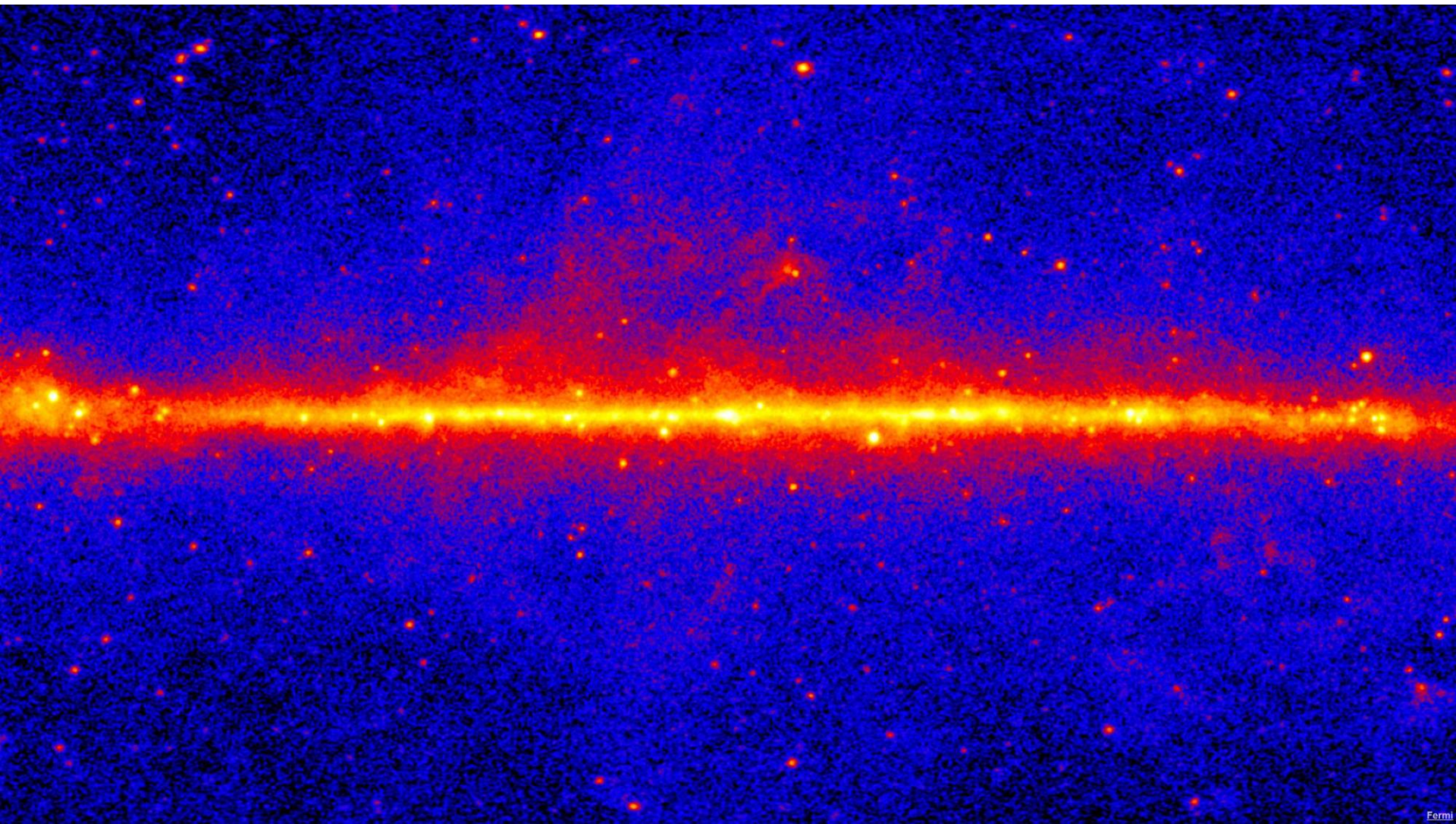
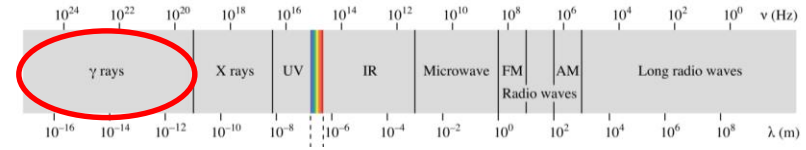
## Optical Spectroscopy in Astronomy

- By looking through a standard telescope, you are observing the night sky at the visible light spectrum





# Gamma Rays



## Black hole binaries in disks of plasma are sources of gamma rays

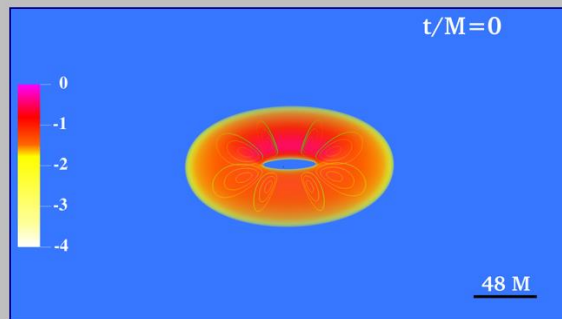


Fig. 2-1: Initial Configuration

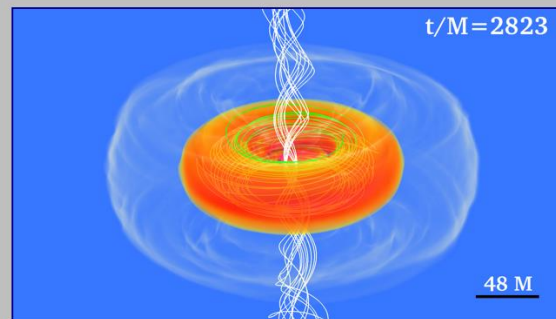


Fig. 2-2: Twin jets rise from the inspiraling black holes

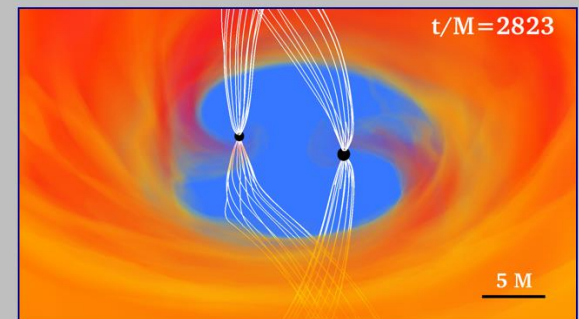


Fig. 2-3: Zoom in to central cavity

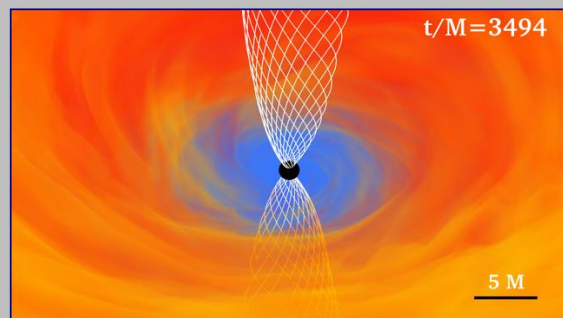


Fig. 2-4: Post merger black hole and magnetic field

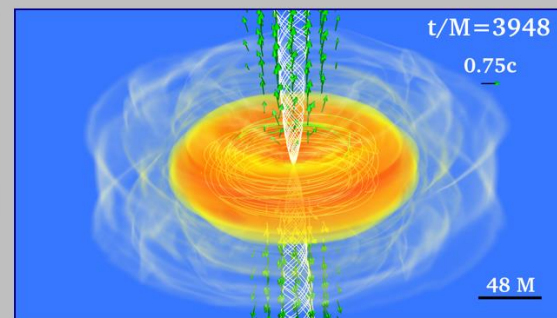
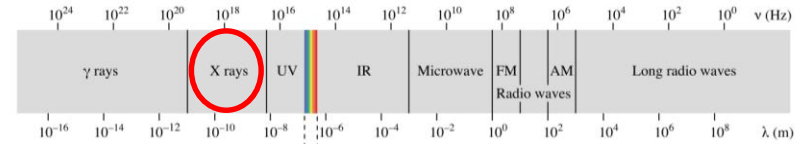


Fig. 2-5: Quasistationary jet outflow



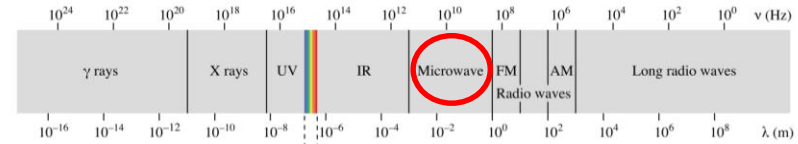
# X Rays



X-ray sources include stars, supernova, gaseous shells ejected during a violent explosion of a dying star, and synchrotron radiation



# Microwave

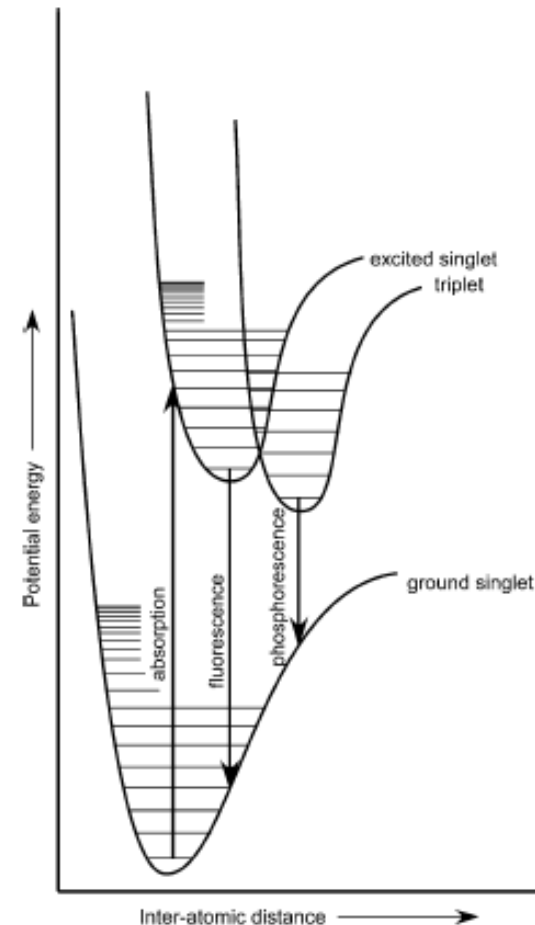
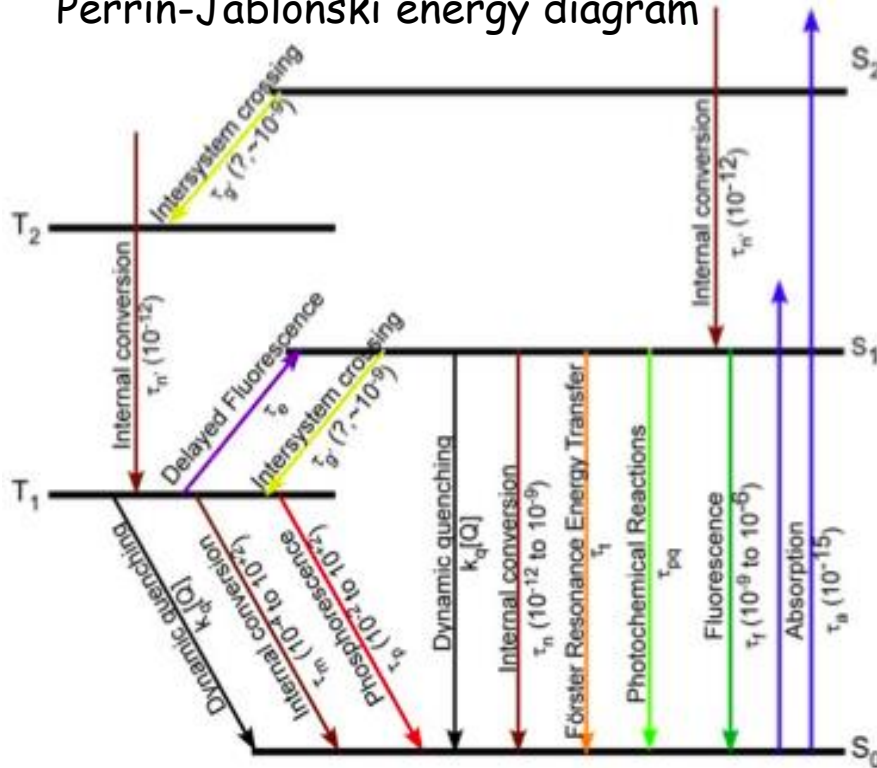


Cosmic microwave background radiation emitted from the big bang and inflation

# Luminescence: Emission of light from any substance

- **Fluorescence:** transition from excited state to ground state is **fast** (~ns – ms range)
- **Phosphorescence:** transition from excited state to ground state is **slow** (~s – ks range)

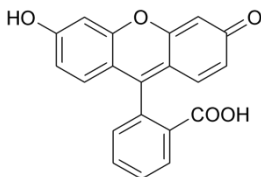
Perrin-Jablonski energy diagram



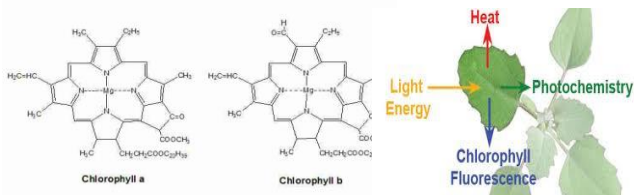


# Types of Fluorescent Molecules

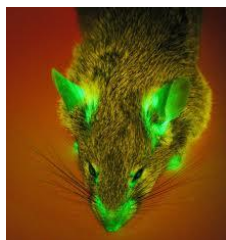
## Synthetic Organic: Fluorescein



## Naturally Occuring:



## Fluorescent Proteins:



Green Fluorescent Protein

## Semiconductor Nanocrystal:

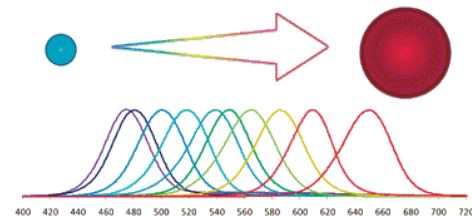
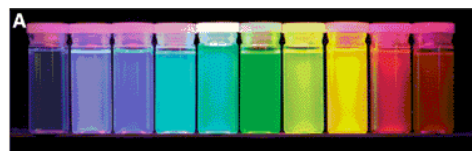
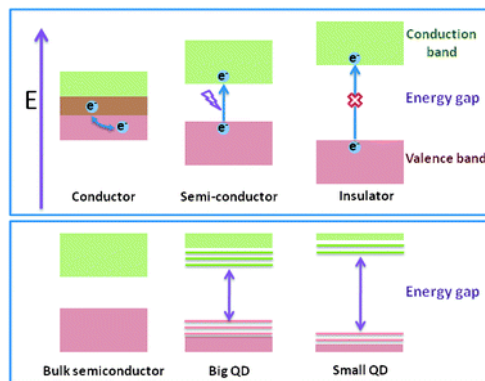
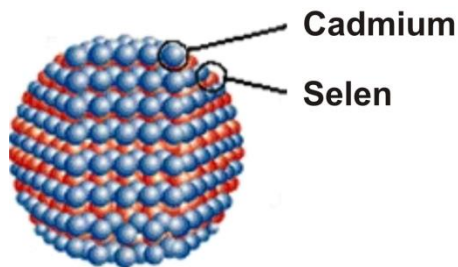
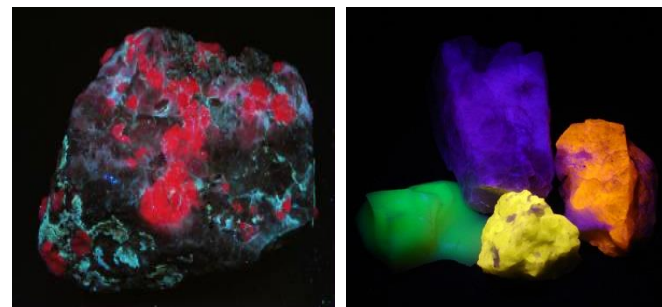


Image from Zrazhevskiy et al. 2010

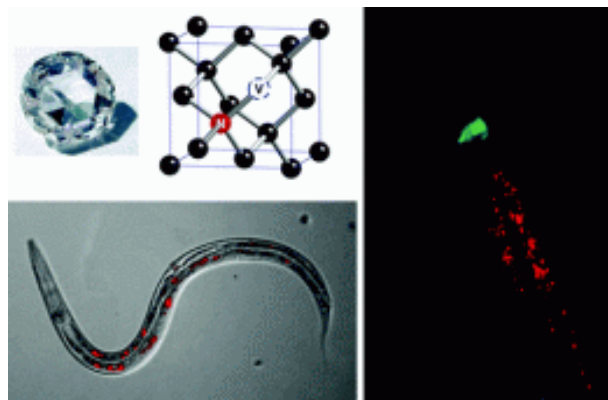
## Crystals:



Ruby and assorted minerals

From mineralman.net

## Fluorescent Nanodiamonds



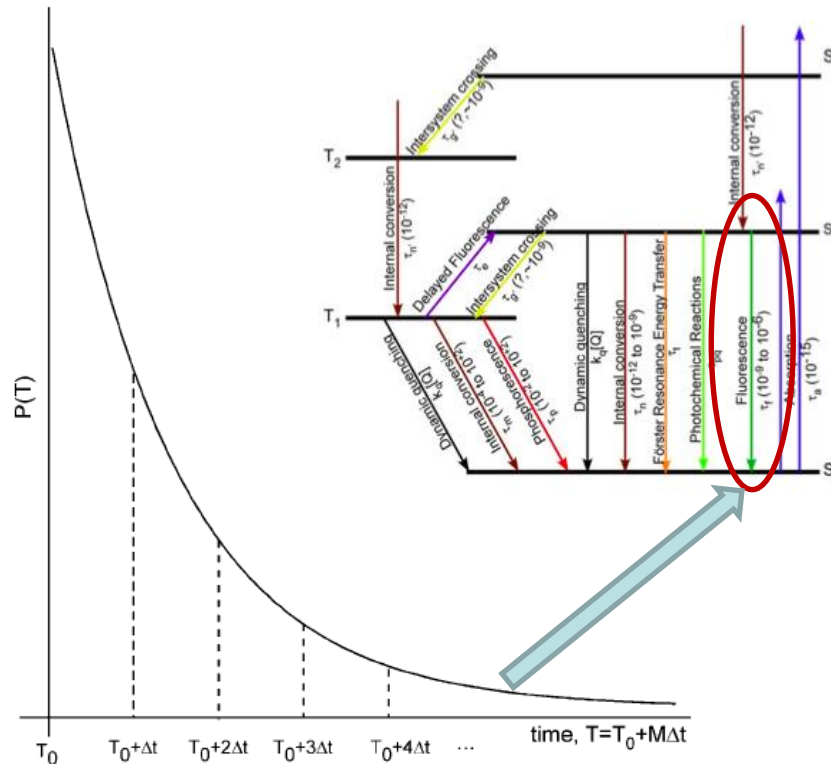
Nano Lett., 2010, 10 (9), pp 3692-3699. DOI: 10.1021/nl1021909



# Time-Dependent Fluorescence: Fluorescence Lifetime

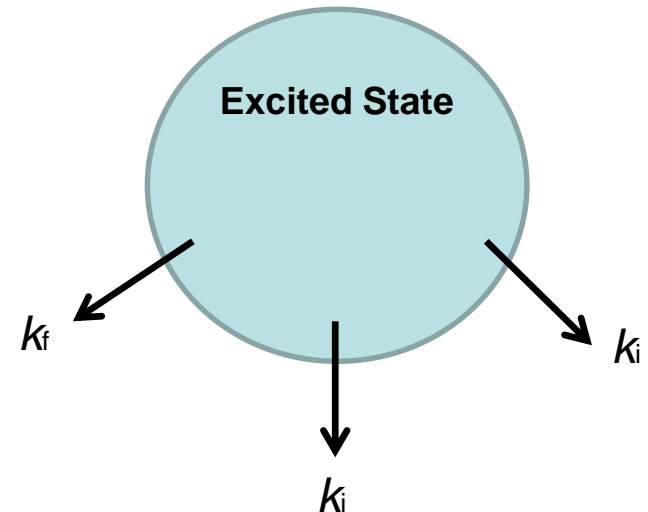
**Fluorescence Lifetime:** The average amount of time a molecule stays in excited state

Probability of being in the excited state



$k_f$  = rate constant for leaving excited state while emitting a photon

$k_i$  = rate constant for leaving excited state through other means (ie. Dynamic quenching, Energy Transfer, etc)



Fluorescence Lifetime: 
$$\tau = \sum_i \frac{1}{k_i}$$

**Lifetime is sensitive to other decaying pathways present!**

## Measuring the Depletion of the excited state

$$\left[ \# x^* \right] = \left[ \# x_o^* \right] e^{-(k_F + k_t)t}$$

$$\left[ \# x^* \right] (k_F) = \text{Intensity that you measure}$$

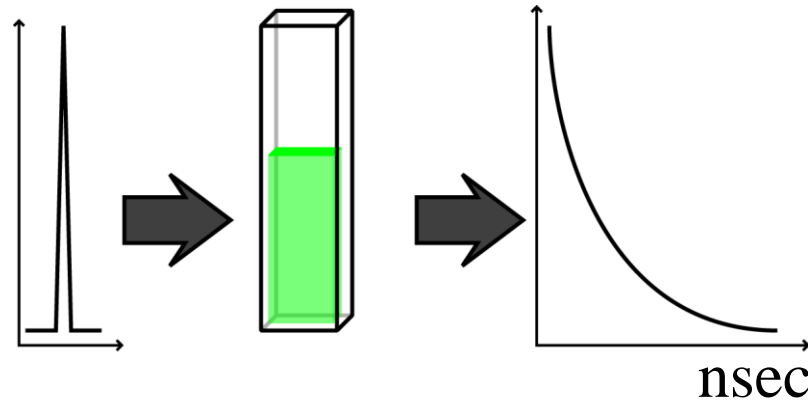


$K_F$  is rate constant of fluorescence

Intensity measured is proportional to the # of molecules in the excited state!



# Measuring Lifetime: Time Domain

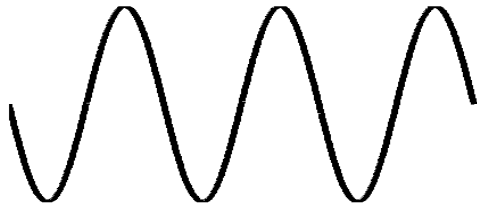


What do you need?

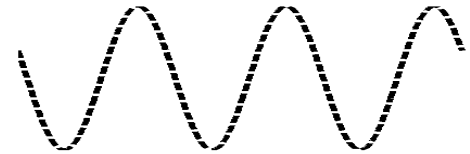
- Collect signal fast enough
- Fitting

# Measuring Lifetime: Frequency Domain

$$E(t) = E_o + E_\omega \cos(\omega_E t + \varphi_E)$$



$$F(t) = F_o + F_\omega \cos(\omega_E t + \varphi_E - \varphi)$$

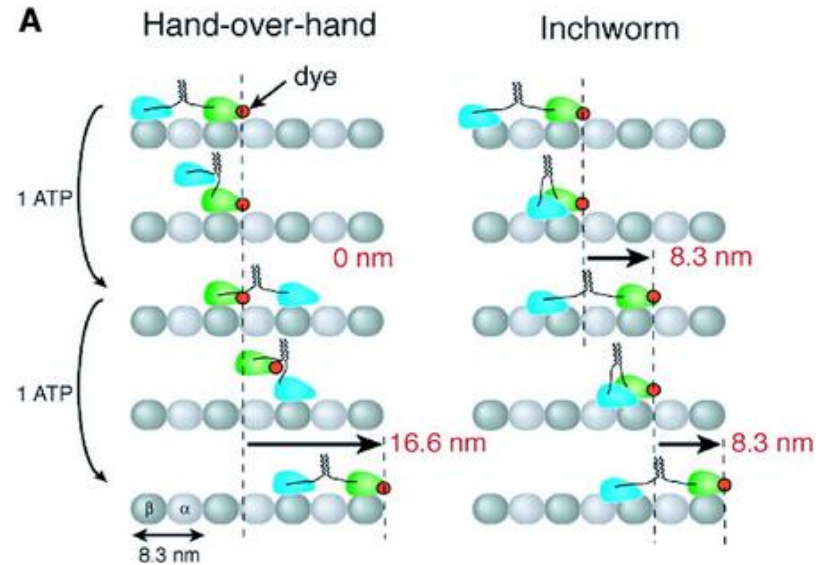
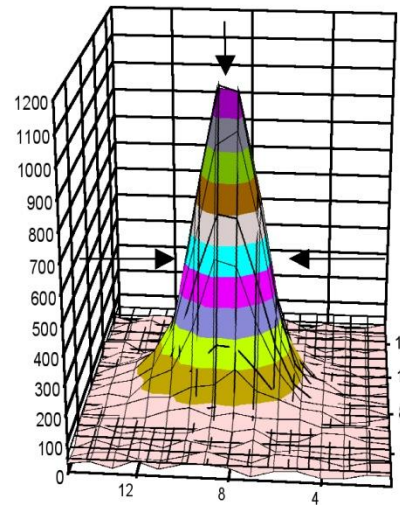
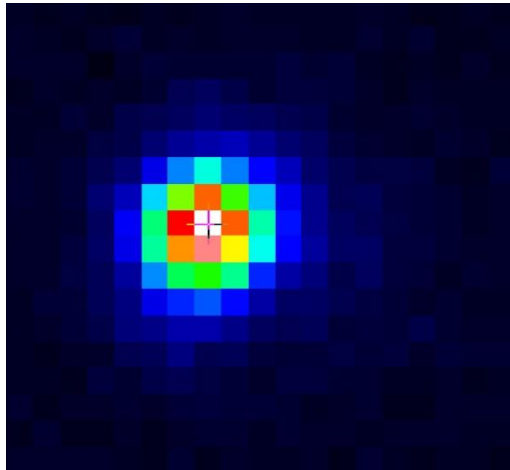


$$\tan(\varphi) = \omega_E \tau_\varphi$$

$$M = \frac{F_\omega / F_o}{E_\omega / E_o} = \frac{1}{\sqrt{1 + (\omega \tau_{Mod})^2}}$$



# Single Molecule Fluorescence Imaging (myosin)



$$\sigma_{\mu_i} = \sqrt{\left(\frac{s_i^2}{N} + \frac{a^2/12}{N} + \frac{8\pi s_i^4 b^2}{a^2 N^2}\right)}$$

Center of the distribution can be determined in ~1.5 nm accuracy if #N is more than 10<sup>4</sup>

